

GE Power & Water Water & Process Technologies

Innovative water treatment solutions for seawater open evaporative systems

1st NACE Jubail Industrial Forum
17 – 19 Oct, 2011

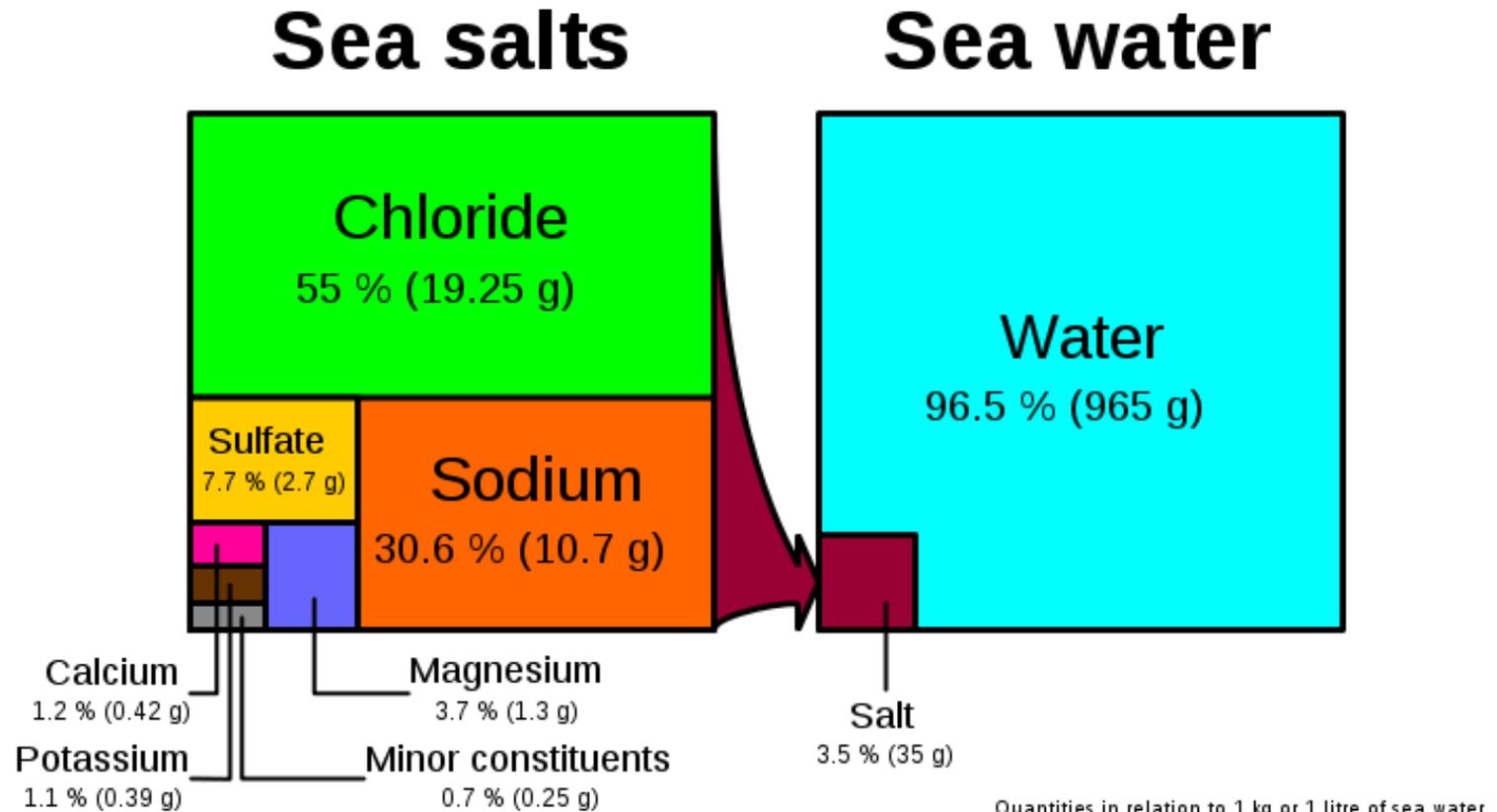
Shereif Alsayed



GE imagination at work

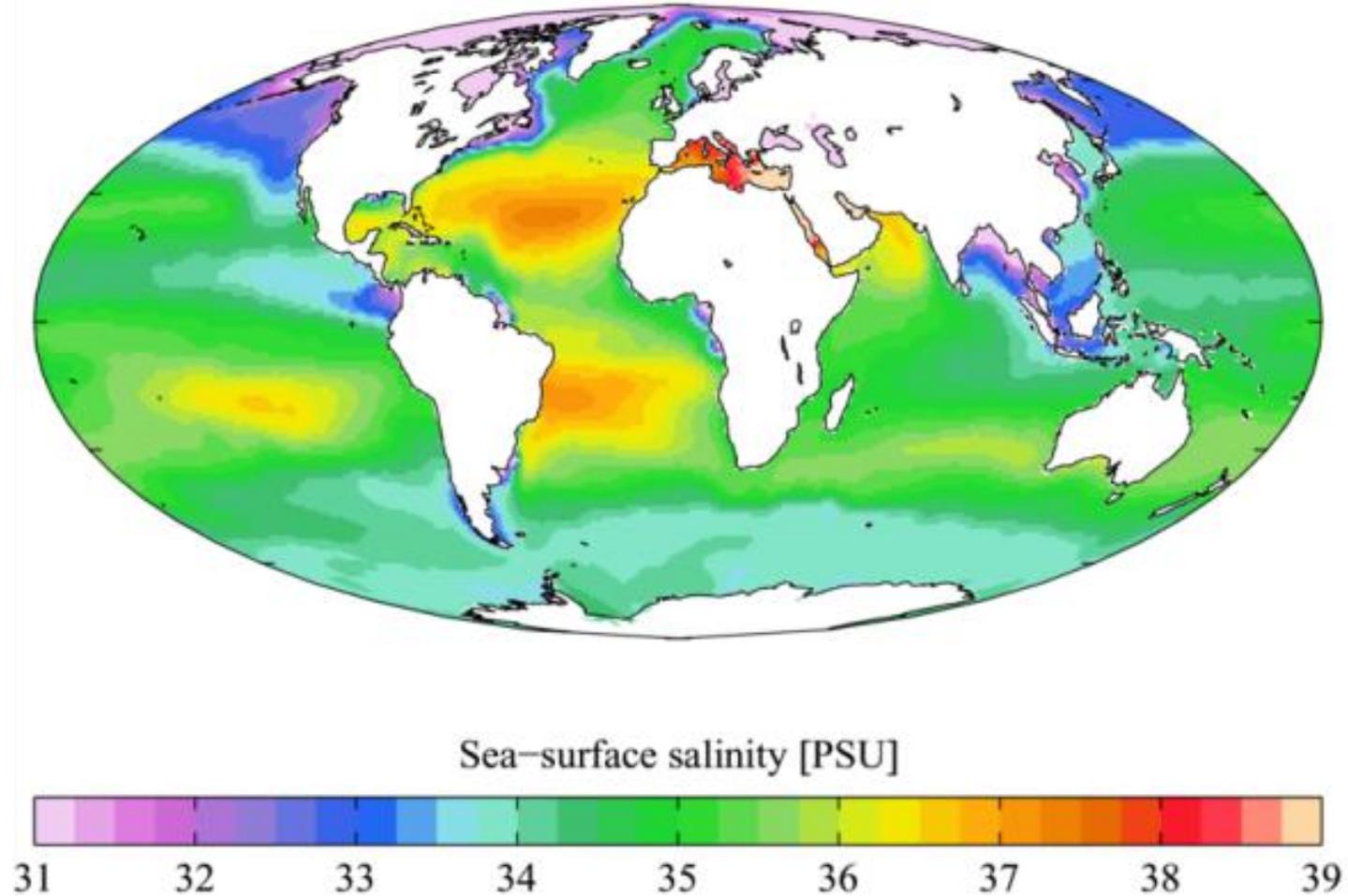
Seawater Properties

Seawater Salinity Composition



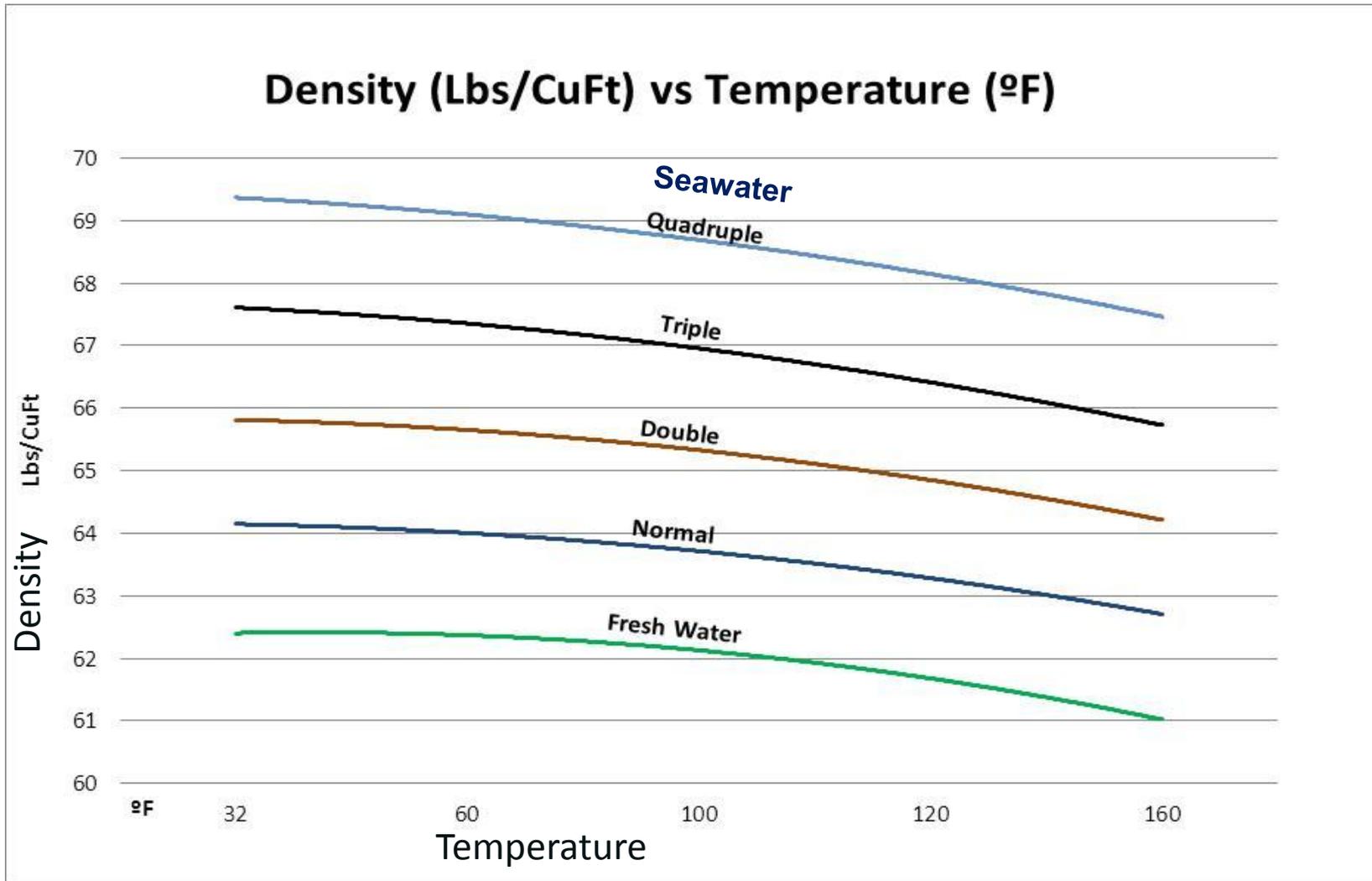
Seawater Properties

Seawater Salinity Distribution

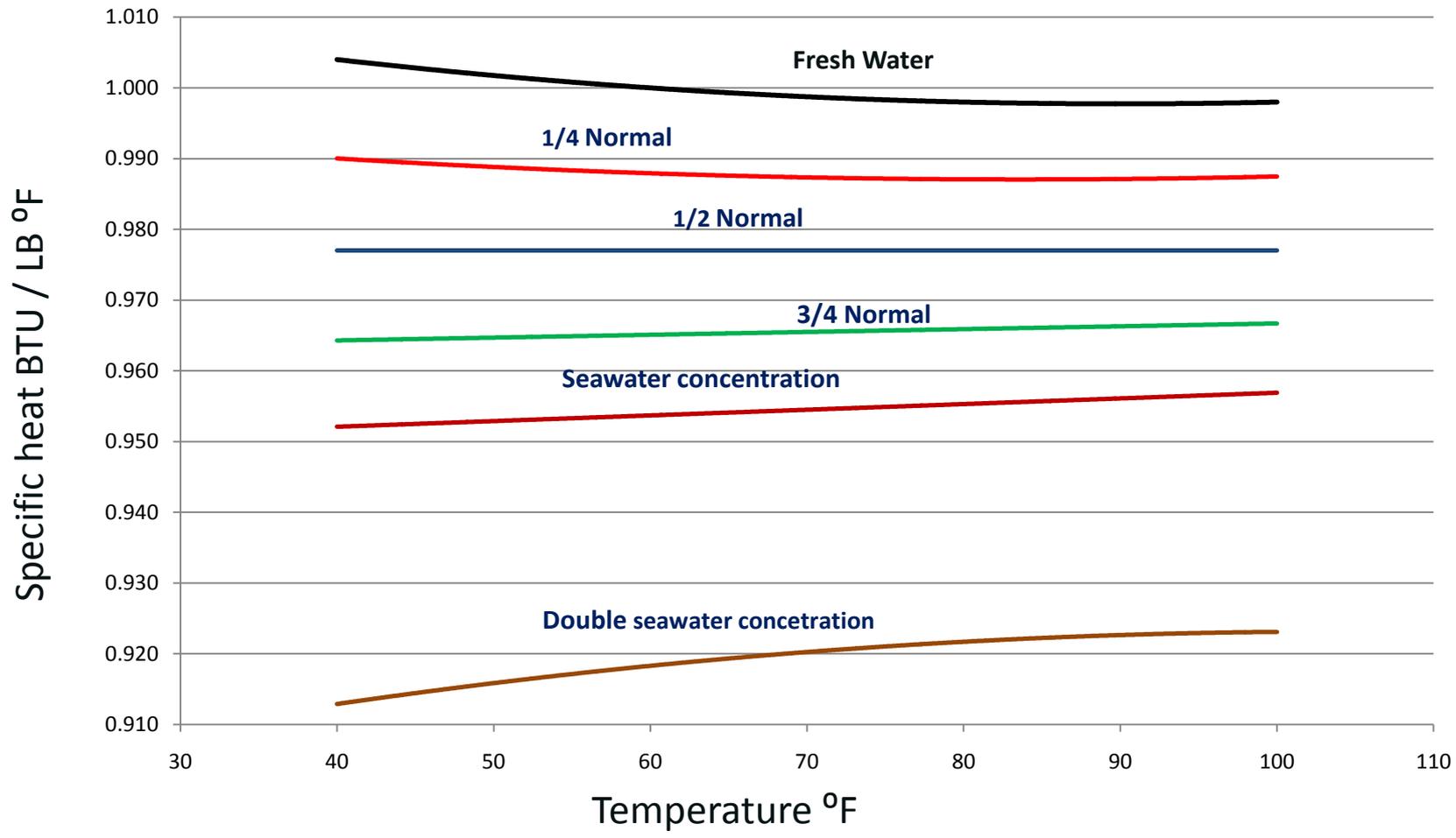


Source: World Ocean Atlas 2005. Retrieved 17 October 2010

Seawater Properties



Seawater Properties



Factors affecting physical properties

PARAMETER	Density	Specific Heat
• Conductivity		
• Temp		
• Depth		
- Desalination plants reject.		
- Vicinity of other plants.		
•		

Seawater Properties

- Effect of lower specific heat

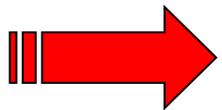
Energy required = heat rejected= Q

$Q = \text{specific heat} \times \text{mass} \times \text{DT}$

$Q(\text{brackish water}) = \text{specific heat} \times \text{mass} \times \text{DT}$

$Q(\text{Seawater}) = \text{specific heat} \times \text{mass} \times \text{DT}$

$Q \text{ brackish water} / Q \text{ seawater} = > 1$



Brackish cooling system is less efficient

Q: Cal ,C_p: Cal/g °C, Mass:g, DT: °C

Q: BTU, C_p: Btu/ Lb °F, Mass: Lb, DT: °F

Seawater Properties

- Effect of higher density

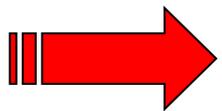
Energy required = heat rejected= Q

Q = specific heat x mass x DT

Q(brackish water) = specific heat x mass x DT

Q(Seawater) = specific heat x mass x DT

Q brackish water / Q seawater = < 1



Seawater cooling system is less efficient

Q: Cal ,C_p: Cal/g °C, Mass:g, DT: °C

Q: BTU, C_p: Btu/ Lb °F, Mass: Lb, DT: °F

Seawater Properties

Loss in energy can be compensated by:

- Increased tower size.
- Adjusting fan horsepower.
- Increased circulation rate.

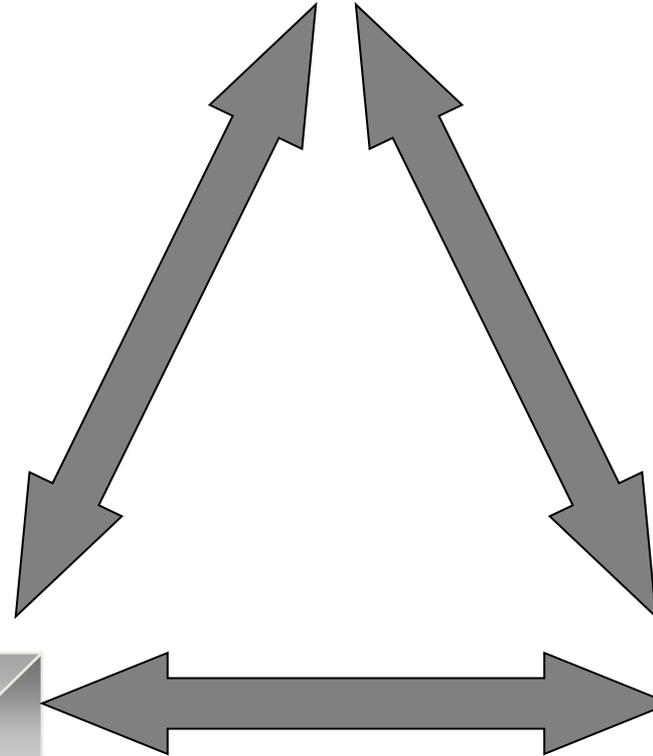
Treatment Concerns



**Corrosion
“Designed out”**

Deposition

Fouling



Scale Formation



Scale Formation

Calcium Carbonate

Scale potential depends on:

- Calcium
- pH control
- Alkalinity
- Temperature
- Ionic strengths

Scale Formation

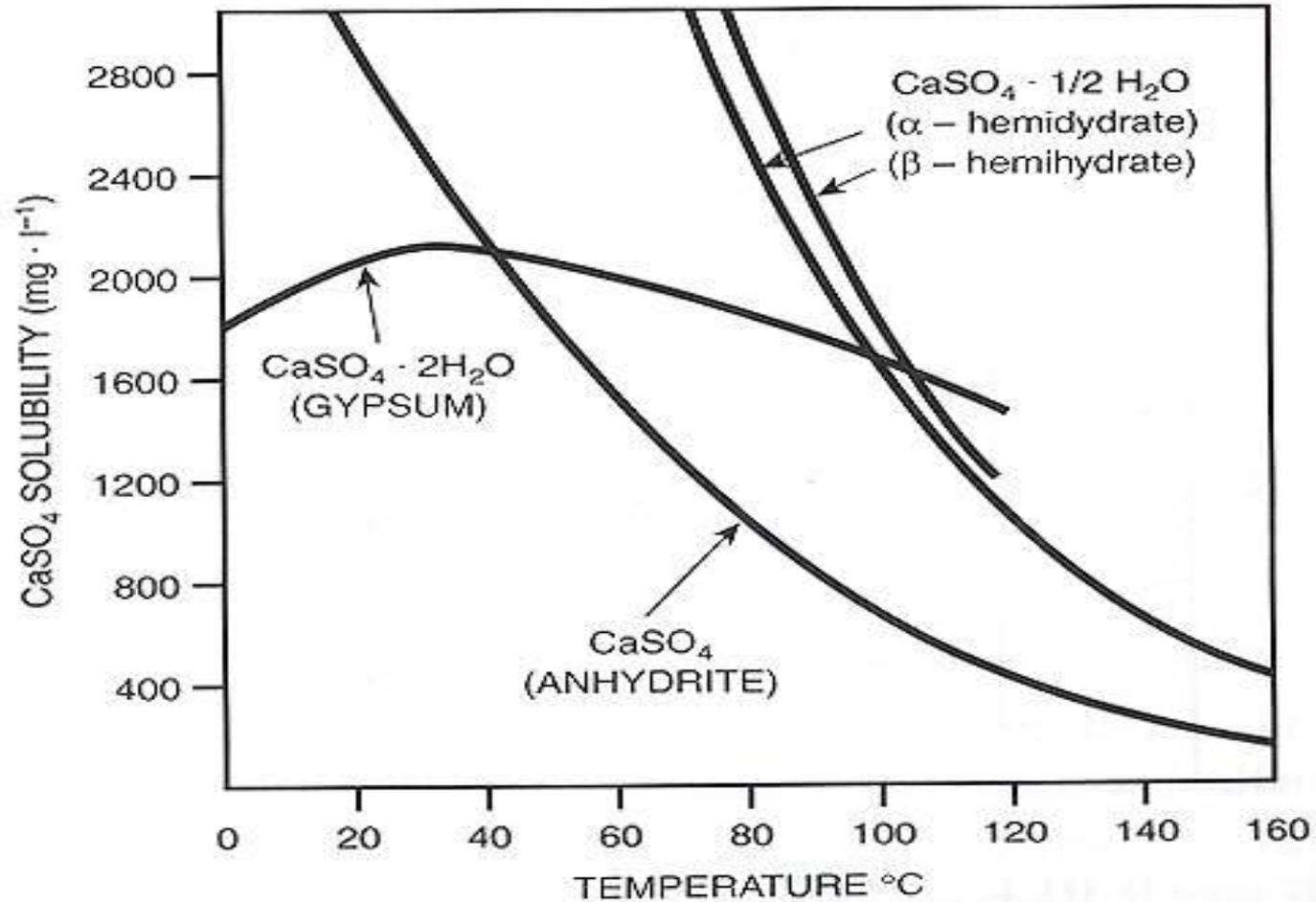
Calcium Sulphate

Scale potential depends on:

- Calcium
- Sulphate
- Ionic strength

Scale Formation

CaSO₄ Solubility



Scale Formation

Silicate

Factors affecting deposition

- Magnesium
- Silicate
- pH

New Technology



Treatment practices overview

- ❑ Typical cycles (1.2 – 1.3).
- ❑ No pH control or adjustment.
- ❑ Continuous chlorination.
- ❑ Material of construction.
 - Sacrificial anode for yellow metallurgy alloys.

Sea Water and Open Evaporative Cooling



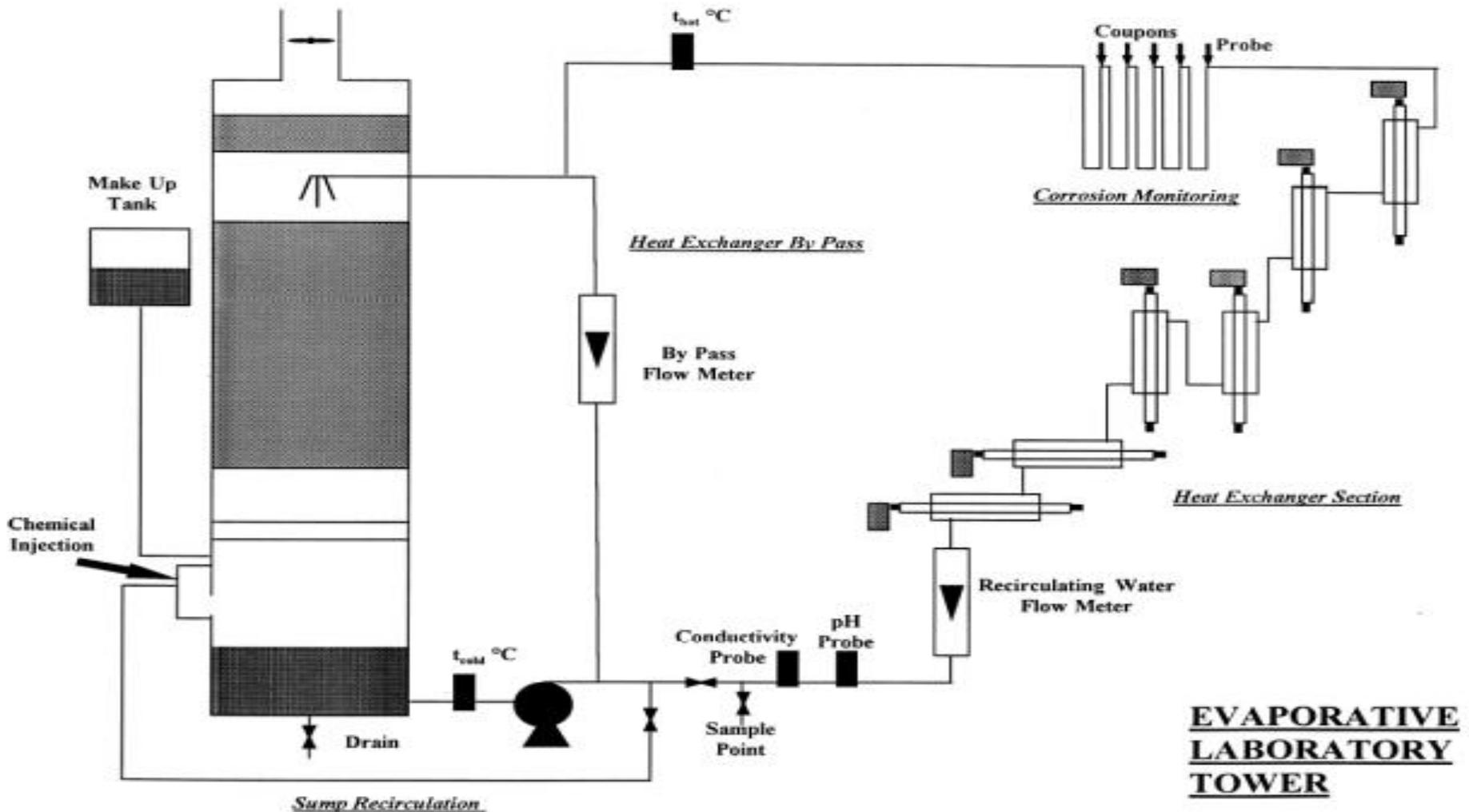
SALT WATER

By John A. Nelson • The Marley Cooling Tower Company • November 5, 1986

“Generally, sea water can be concentrated to approximately 55,000 ppm salinity with no serious scaling problems in the exchange. Higher concentrations are possible but pH control by acid addition probably be required. Two of the major users of sea water cooling towers operate to 55,000 ppm salinity as the upper limit and this procedure has been satisfactory.”

GE technology reached > 80,000 ppm, free pH

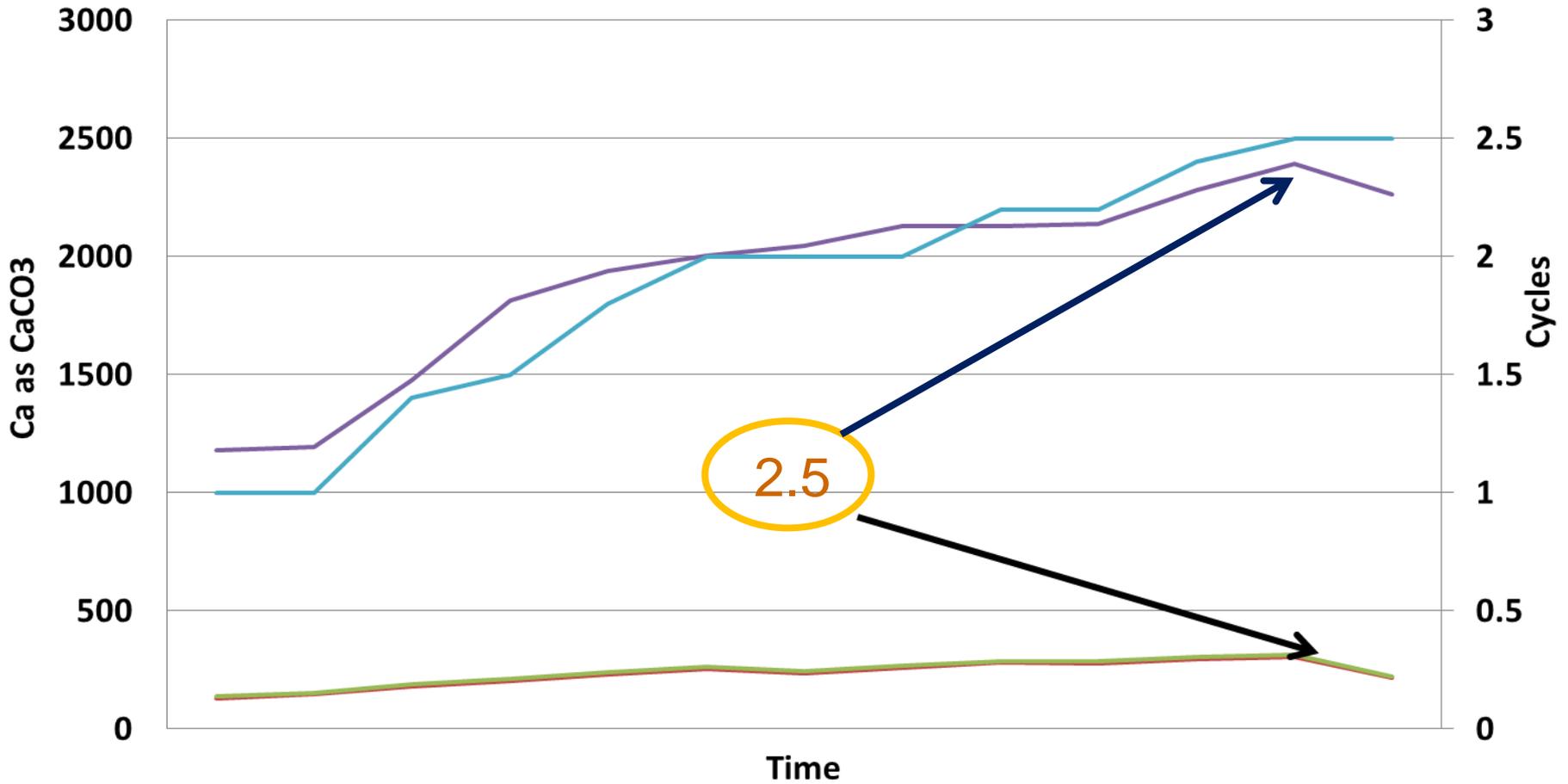
Evaluation of Inhibitors



Operating Conditions

- Coupons: Ti , 70/30 Cu/Ni,
- CRM Probe: 70/30 Cu/Ni
- Heat Exchanger Tubes: Ti, 70/30 Cu/Ni
- Water Temperature: 40 °C
- Estimated Skin Temperature: 52 °C
- Water Velocity: 1.5 m/s

pH, Alkalinity , Ca .vs. Cycles



Ca as CaCO3

pH

Alkalinity as CaCO3

Cycles

Corrosion Coupons



CRM probe corrosion Cu/ Ni 70-30 : 0.9 mpy

Heat exchanger tubes

Titanium Gr2

0.1 mpy

0.2 mpy

0.1 mpy

Cu Ni 70 -30

1.8 mpy

2.8 mpy

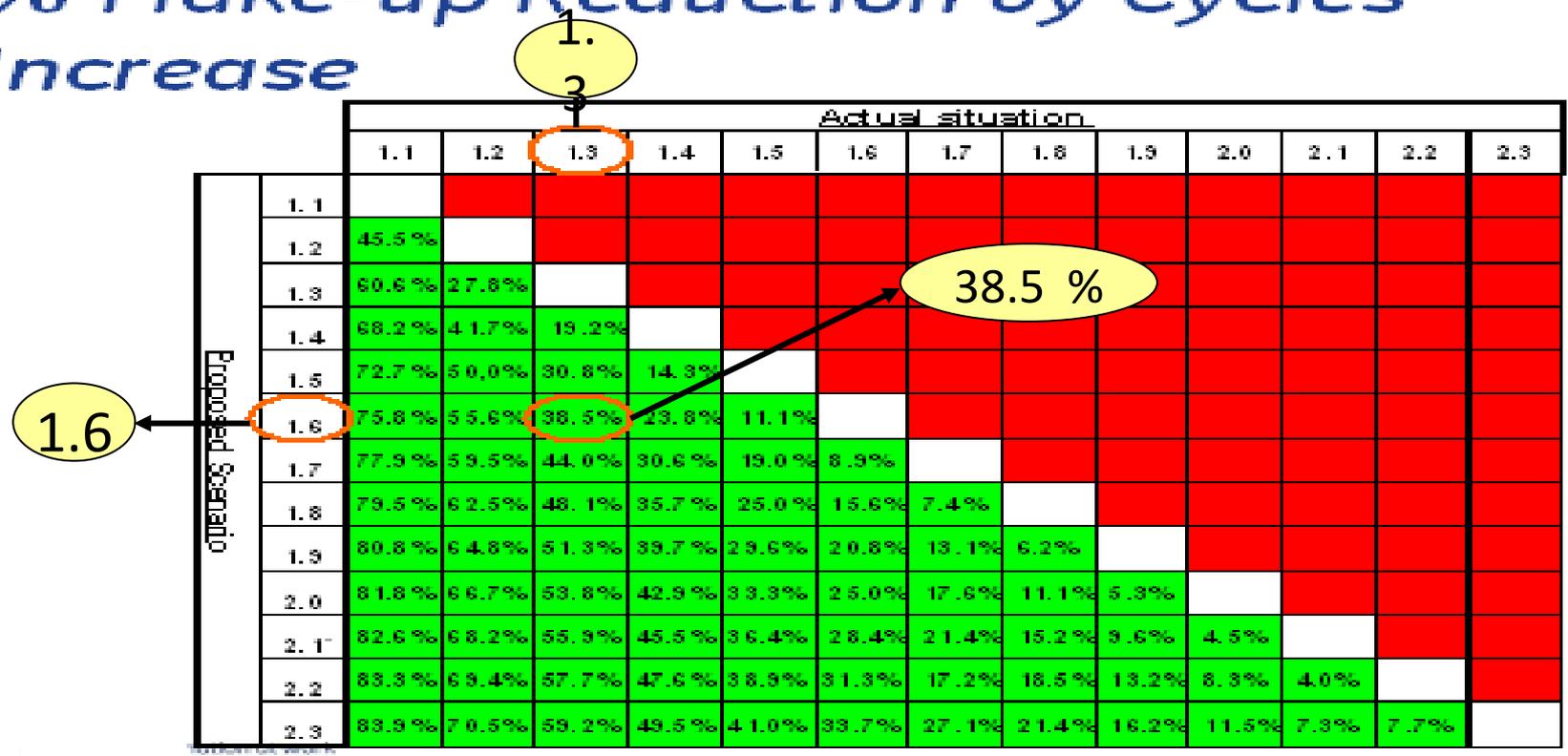
1.7 mpy

Technology Guide

- ❑ Stiff & Davis Max 1.50 @ 52°C (skin)
- ❑ Up to theoretical LSI of 2.5 @ 52°C.
- ❑ Max 2.4 cycles (depending on chemistry).
- ❑ Free pH, pH of 8.4.



% Make-up Reduction by Cycles Increase



- ❑ 38.5 % saving in make up requirement.
- ❑ Reduced energy consumption (make up, blow down pumps)
- ❑ Forgiving program, higher tolerance for upsets and excursions.
- ❑ Compatible with halogenation practices.

Technology Features & Advantages

